The results of this work agree with the enhanced tendency of compound 3c to undergo dimerization under the action of bases.<sup>4</sup>

The characteristics of the compounds synthesized are given below.

2-Benzoylspiro[2.5]octane-1,1-dicarbouitrile (5a). M.p. 114—115 °C (EtOH). IR (KBr), ν/cm<sup>-1</sup>: 2255 (CN), 1680 (CO). <sup>1</sup>H NMR ((CD<sub>3</sub>)<sub>2</sub>CO), δ: 1.4—2.3 (m, 10 H); 4.00 (s, 1 H, H(2)); 7.62 (t, 2 H); 7.74 (t, 1 H); 8.11 (d, 2 H) (all H<sub>ph</sub>).

**2-Benzoyl-6-(***tert*-butyl)spiro[2.5]octane-1,1-dicarbonitrile (5b). M.p. 134–135 °C (EtOH). <sup>1</sup>H NMR ((CD<sub>3</sub>)<sub>2</sub>CO),  $\delta$ : 0.93 (s, 9 H; But); 1.2–2.1 (m, 8 H); 2.40 (m, 1 H, H(6)); 3.95 (s, 1 H; H(2)); 7.60 (t, 2 H); 7.73 (t, 1 H); 8.05 (d, 2 H) (all H<sub>Pli</sub>).

6'-Amino-3',3'a,4',5'-tetrahydrospiro[cyclopentane-1,4'(2'H)-indene]-5',5',7'-tricarbonitrile (6). M.p. 182–184 °C (EtOH) (cf. Ref. 4: m.p. 184–186 °C). <sup>1</sup>H NMR ((CD<sub>3</sub>)<sub>2</sub>CO), δ: 1.4-2.2 (m, 10 H); 2.46 (m, 2 H, 2 H(2')); 3.15 (m, 1 H; H(3'a)); 5.63 (m, 1 H; H(1')); 6.7 (br.s, 2 H, NH<sub>2</sub>).

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## A new one-step method for the synthesis of 1-(alkoxy-NNO-azoxy)-2-phenylethenes from di(alkoxy-NNO-azoxy)methanes

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In a single publication, where (alkoxy-NNO-azoxy) olefins have been mentioned, I-(methoxy-NNO-azoxy)-2-phenylethene (1a) was obtained by the pyrolysis of compound 2a synthesized in four steps from bisazoxymethane 3a or ketone 4. No reaction conditions, yields, and properties of the products were reported.

We attempted to synthesize an intermediate product 2a by the direct C-benzylation of compound 3a. However, the reaction unexpectedly furnished the target olefin 1a in 46% yield. The ethyl homologue 1b was synthesized similarly in 42% yield. Compounds 2a and 2b are formed in the first step. They predominate, according to TLC, over 1a,b and 3a,b during the first 10 min and disappear completely in 1 h.

1-(Methoxy-NNO-azoxy)-2-phenylethene (1a). Aqueous 45% NaOH (2.6 mL, 0.06 mol) was added with stirring and cooling with tap water to a solution of compound  $3a^{-2}$  (3.28 g. 0.02 mol), PhCH<sub>2</sub>Cl (5.06 g, 0.04 mol), and Et<sub>4</sub>NBr (0.42 g, 0.002 mol) in DMSO (10 mL). The reaction was monitored by TLC on Silufol (PhH—EtOAc, 3:1):  $R_f$  for 1a, 0.09; for 2a, 0.29; and for 3a, 0.40. One hour later, the reaction mixture was diluted with water, acidified with HCl, and extracted with

**Reagents:** a. 1) PhCHO, McONa, 2)  $Ac_2O$ , 3)  $Et_3N$ , 4) [H]; b. 1) NO, McONa, 2)  $H_2O/OH^-$ , 3)  $AgNO_3$ , 4) McI.

CHCl<sub>3</sub> (3×20 mL). The extract was washed with 10%  $Na_2CO_3$  (10 mL), a saturated solution of NaCl (2×20 mL), and water (20 mL), and concentrated *in vacuo*. The residue (3.35 g) was

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$$\beta$$
 O N—O(CH<sub>2</sub>)<sub>n</sub>Me  
3: R = Me (a), Et (b) 1a,b (n = 0 (a), 1(b))

Reagents: a. PhCH<sub>2</sub>Cl, NaOH, Et<sub>4</sub>NBr or TEBAC, DMSO; b. OH<sup>-</sup>

flash-chromatographed on Silpearl (20 g) using CHCl<sub>3</sub> as the eluent. The fraction containing 1a (2.50 g) was recrystallized from Et<sub>2</sub>O to obtain 1a (1.65 g, 46%) with m.p. 80-81 °C. UV (water),  $\lambda$ /nm ( $\epsilon$ ): 200 (13700), 232 (8000), 291 (22700).

IR (film),  $v/cm^{-1}$ : 3100, 2965, 1480 (N<sub>2</sub>O<sub>2</sub>), 1310, 1220, 1090, 1050, 995, 885, 865, 850, 780, 710. <sup>1</sup>H NMR (cf. Ref. 1),  $\delta$  (10% in DMSO-d<sub>6</sub>): 4.12 (s, 3 H, CH<sub>3</sub>); 7.42—7.46 (m, 3 H, m-H and p-H); 7.68 (d, 1 H,  $\alpha$ -H,  $J_{\alpha,\beta}$  = 13.7 Hz); 7.75—7.78 (m, 2 H, o-H); 8.07 (d, 1 H,  $\beta$ -H); (10% in CCl<sub>4</sub>): 4.04 (s, 3 H, CH<sub>3</sub>); 7.29—7.41 (m, 6 H, H arom. +  $\alpha$ -H); 7.60 (d, 1 H,  $\beta$ -H,  $J_{\alpha,\beta}$  = 13.3 Hz).

1-(Ethoxy-NNO-zzoxy)-2-phenylethene (1b) was obtained similarly from 2b, <sup>2</sup> with benzyltriethylammonium chloride instead of Et<sub>4</sub>NBr;  $R_f$  for 1b, 0.12; for 2b, 0.32; and for 3b, 0.44. The yield of 1b was 1.62 g (42%), m.p. 46–47 °C (CHCl<sub>3</sub>-hexane). <sup>1</sup>H NMR (10% in DMSO-d<sub>6</sub>), δ: 1.34 (t, 3 H, CH<sub>3</sub>, J = 7.1 Hz); 4.40 (q, 2 H, CH<sub>2</sub>); 7.40–7.44 (m, 3 H, m-H + p-H); 7.66 (d, 1 H, α-H, J = 13.5 Hz); 7.73–7.76 (m, 2 H, o-H); 8.07 (d, 1 H, β-H). MS (EI, 70 eV), m/z ( $I_{rel}$  (%)): 192 [M]<sup>+</sup> (43), 163 [M—Et]<sup>+</sup> (17), 135 (17), 133 [M—Et—NO]<sup>+</sup> (89), 117 (16), 116 (10), 106 (12), 105 (19), 104 (47), 103 [PhCH=CH]<sup>+</sup> (68), 101 (23), 91 [PhCH<sub>2</sub>]<sup>+</sup> (20), 90 [PhCH]<sup>+</sup> (12), 88 (11), 80 (20), 78 (29), 77 [Ph]<sup>+</sup> (100), 63 (14), 51 (45), 49 (22), 39 (12).

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## Polyfluorinated enamines. New methods for the synthesis of 5-trifluoromethyluracil

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5-Trifluoromethyluracil (1) is known to possess high anticancer and antiviral activities. However, the methods for its preparation<sup>2-6</sup> have several disadvantages, such as the multi-stage character, difficultly accessible starting reagents, 3,4 the use of organomercury derivatives, and low yields of the target product. 2,5,6

We synthesized compound 1 by two new methods, viz., by the reaction of cis, trans-3-dimethylamino-2-trifluoromethacryloyl fluoride<sup>7,8</sup> (2) with urea and by cyclization with partial hydrolysis of N-(3,3,3-trifluoro-2-trifluoromethylprop-1-enyl)urea<sup>9</sup> (3). Compounds 2

and 3 were obtained in a few steps from octafluoroisobutene, which is a large-scale by-product of the industrial production of fluoroplastics.